126. Select the total number of substance which is/are given as their most stable allotrope of corresponding element : Graphite, Ozone, Black phosphorus, Rhombic sulphur, White tin

127. What is the change in oxidation number of nitrogen when very dil. HNO_3 reacts with Zn metal ?

128. Total number of substance which contain hexagonal planar rings in their structures graphite, $(BN)_x$, $B_3N_3H_6$, C_6H_6, B_2H_6, $H_3P_3O_9$:

129. Total number of moles of hydrochloric acid react with one mole of borax to convert all borons to boric acid :

130. How many moles of PH_3 gas is produced when 1 mole of calcium phosphide reacts with excess of water ?

131. Number of oxides of nitrogen which produce HNO_3 when dissolved in water :
N_2O, NO, N_2O_3, NO_2, N_2O_4, N_2O_5

132. Maximum number of 'O' atoms are bounded with each Si in SiO_2 :

133. Maximum number of halogen are possible in uncharge interhalogen compound :

134. Number of moles of NaOH required for complete neutralisation of H^+ in solution which is formed by hydrolysis of 1 mole of PCl_5.

135. In the compound $Na_2\left[B_4O_5(OH)_4\right]\cdot 8H_2O$, if the

    (i)   number of $B-O-B$ bonds is $'x'$

    (ii)   number of $B-B$ bonds is $'y'$

    (iii)   number of sp^2 hybridized B atoms is $'z'$

Calculate the value of $x+y+z$.